# Ranvet Pty Ltd

Chemwatch: 4787-91

Version No: 7.1 Safety Data Sheet according to Work Health and Safety Regulations (Hazardous Chemicals) 2023 and ADG requirements Chemwatch Hazard Alert Code: 2

Issue Date: **03/10/2023** Print Date: **08/08/2024** L.GHS.AUS.EN.E

# SECTION 1 Identification of the substance / mixture and of the company / undertaking

Product Identifier	
Ranvet's Hoof Food	
Not Applicable	
Not Available	
Not Applicable	
Not Available	

# Relevant identified uses of the substance or mixture and uses advised against

Relevant identified uses Hoof growth promoter containing BIOTIN and other essential nutrients.

# Details of the manufacturer or supplier of the safety data sheet

Ranvet Pty Ltd	
10-12 Green Street Banksmeadow NSW 2019 Australia	
+61 2 9666 1744	
+61 2 9666 1755	
https://www.ranvet.com.au/other msds.htm	
info@ranvet.com.au	

### Emergency telephone number

Association / Organisation	Ranvet Pty Ltd
Emergency telephone numbers	+61 417 580 980
Other emergency telephone numbers	Not Available

### **SECTION 2 Hazards identification**

# Classification of the substance or mixture

HAZARDOUS CHEMICAL. NON-DANGEROUS GOODS. According to the WHS Regulations and the ADG Code.

# Chemwatch Hazard Ratings

	Min	Max	
Flammability	1		
Toxicity	1		0 = Minimum
Body Contact	2		1 = Low
Reactivity	1		2 = Moderate
Chronic	2		3 = High 4 = Extreme

Poisons Schedule	Not Applicable
Classification <sup>[1]</sup>	Skin Corrosion/Irritation Category 2, Sensitisation (Skin) Category 1, Serious Eye Damage/Eye Irritation Category 2A, Specific Target Organ Toxicity - Single Exposure (Respiratory Tract Irritation) Category 3, Specific Target Organ Toxicity - Repeated Exposure Category 2
Legend:	1. Classified by Chemwatch; 2. Classification drawn from HCIS; 3. Classification drawn from Regulation (EU) No 1272/2008 - Annex VI

### Label elements

Hazard pictogram(s)	

Signal word Warning

# Hazard statement(s)

.,	
H315	Causes skin irritation.
H317	May cause an allergic skin reaction.
H319	Causes serious eye irritation.
H335	May cause respiratory irritation.

H373	May cause damage to organs through prolonged or repeated exposure.
autionary statement(s) Pr	evention
P260	Do not breathe dust/fume.
P271	Use only outdoors or in a well-ventilated area.
P280	Wear protective gloves, protective clothing, eye protection and face protection.
P264	Wash all exposed external body areas thoroughly after handling.
P272	Contaminated work clothing should not be allowed out of the workplace.
1212	
cautionary statement(s) Re	sponse
cautionary statement(s) Re P302+P352	IF ON SKIN: Wash with plenty of water and soap.
ecautionary statement(s) Re P302+P352 P305+P351+P338	IF ON SKIN: Wash with plenty of water and soap. IF IN EYES: Rinse cautiously with water for several minutes. Remove contact lenses, if present and easy to do. Continue rinsing.
ecautionary statement(s) Re P302+P352 P305+P351+P338 P312	IF ON SKIN: Wash with plenty of water and soap. IF IN EYES: Rinse cautiously with water for several minutes. Remove contact lenses, if present and easy to do. Continue rinsing. Call a POISON CENTER/doctor/physician/first aider/if you feel unwell.
ecautionary statement(s) Re P302+P352 P305+P351+P338 P312 P333+P313	IF ON SKIN: Wash with plenty of water and soap. IF ON SKIN: Wash with plenty of water and soap. IF IN EYES: Rinse cautiously with water for several minutes. Remove contact lenses, if present and easy to do. Continue rinsing. Call a POISON CENTER/doctor/physician/first aider/if you feel unwell. If skin irritation or rash occurs: Get medical advice/attention.

P405	Store locked up.	
P403+P233	Store in a well-ventilated place. Keep container tightly closed.	

### Precautionary statement(s) Disposal

P501

Dispose of contents/container to authorised hazardous or special waste collection point in accordance with any local regulation.

### **SECTION 3 Composition / information on ingredients**

### Substances

See section below for composition of Mixtures

### Mixtures

CAS No	%[weight]	Name
63-68-3	<30	methionine
Not Available	<30	vitamins and nutrients
Not Available	balance Ingredients determined not to be hazardous	
Legend:	nd: 1. Classified by Chemwatch; 2. Classification drawn from HCIS; 3. Classification drawn from Regulation (EU) No 1272/2008 - Annex VI; 4. Classification drawn from C&L * EU IOELVs available	

### **SECTION 4 First aid measures**

#### Description of first aid measures If this product comes in contact with the eves: Wash out immediately with fresh running water. Ensure complete irrigation of the eye by keeping eyelids apart and away from eye and moving the eyelids by occasionally lifting the Eye Contact upper and lower lids. Seek medical attention without delay; if pain persists or recurs seek medical attention. Removal of contact lenses after an eye injury should only be undertaken by skilled personnel. If skin contact occurs: Immediately remove all contaminated clothing, including footwear. Skin Contact Flush skin and hair with running water (and soap if available). Seek medical attention in event of irritation. If fumes, aerosols or combustion products are inhaled remove from contaminated area. Inhalation Other measures are usually unnecessary. If swallowed do NOT induce vomiting. If vomiting occurs, lean patient forward or place on left side (head-down position, if possible) to maintain open airway and prevent aspiration. Ingestion Observe the patient carefully. • Never give liquid to a person showing signs of being sleepy or with reduced awareness; i.e. becoming unconscious. Give water to rinse out mouth, then provide liquid slowly and as much as casualty can comfortably drink.

Indication of any immediate medical attention and special treatment needed

Seek medical advice

Treat symptomatically.

# **SECTION 5 Firefighting measures**

### Extinguishing media

- There is no restriction on the type of extinguisher which may be used.
- Use extinguishing media suitable for surrounding area.

Fire Incompatibility	Avoid contamination with oxidising agents i.e. nitrates, oxidising acids, chlorine bleaches, pool chlorine etc. as ignition may result
Advice for firefighters	
Fire Fighting	<ul> <li>Alert Fire Brigade and tell them location and nature of hazard.</li> <li>Wear breathing apparatus plus protective gloves in the event of a fire.</li> <li>Prevent, by any means available, spillage from entering drains or water courses.</li> <li>Use fire fighting procedures suitable for surrounding area.</li> <li>DO NOT approach containers suspected to be hot.</li> <li>Cool fire exposed containers with water spray from a protected location.</li> <li>If safe to do so, remove containers from path of fire.</li> <li>Equipment should be thoroughly decontaminated after use.</li> </ul>
Fire/Explosion Hazard	<ul> <li>Combustible solid which burns but propagates flame with difficulty; it is estimated that most organic dusts are combustible (circa 70%) - according to the circumstances under which the combustion process occurs, such materials may cause fires and / or dust explosions.</li> <li>Organic powders when thely divide over a range of concentrations regardless of particulate size or shape and suspended in air or some other oxidizing medium may form explosive dust-air mixtures and result in a fire or dust septosion (including secondary explosions).</li> <li>Avoid generating dust, particularly clouds of dust in a confined or unventilated space as dusts may form a explosive mixture with air, and any source of gnithon, i.e. fame or spark, will cause fire or explosion. Dust clouds generated by the fine grinding of the solid are a particular hazard; accumulations of fine dust (420 micron or less) may burn rapidly and fercely if gnited - particles exceeding this limit will generally not form flammable dust clouds; once initiated, however, larger particles up to 1400 microns diameter will contribute to the propagation of an explosive limit (LEL) and upper explosive limit (UEL) are applicable for a clouds but only the LEL is of practical use; - this is because of the inherent difficulty of achieving horogeneous dust clouds at high temperatures (for dusts the LEL is of practical use; - this in because of the inherent difficulty of achieving horogeneous dust clouds at high temperatures (for dusts the LEL is of practical use; - this in because of the inherent difficulty of achieving horogeneous dust clouds at high temperatures (for dusts the LEL is of practical use; - this in because of the inherent difficulty or dusts or dusts.</li> <li>A dust explosion may release of large quantities of gaseous products; this in turn creates a subsequent pressure rise of explosive force capable of damaging plant and buildings and injuring people.</li> <li>Usually the initial or primary explosion takes place in a confi</li></ul>
	May emit corrosive fumes.

# **SECTION 6** Accidental release measures

Personal precautions, protective equipment and emergency procedures See section 8

# **Environmental precautions**

See section 12

# Methods and material for containment and cleaning up

wethous and material for cont	anment and cleaning up
Minor Spills	<ul> <li>Remove all ignition sources.</li> <li>Clean up all spills immediately.</li> <li>Avoid contact with skin and eyes.</li> <li>Control personal contact with the substance, by using protective equipment.</li> <li>Use dry clean up procedures and avoid generating dust.</li> <li>Place in a suitable, labelled container for waste disposal.</li> </ul>
Major Spills	<ul> <li>Moderate hazard.</li> <li>CAUTION: Advise personnel in area.</li> <li>Alert Emergency Services and tell them location and nature of hazard.</li> <li>Control personal contact by wearing protective clothing.</li> <li>Prevent, by any means available, spillage from entering drains or water courses.</li> <li>Recover product wherever possible.</li> <li>IF DRY: Use dry clean up procedures and avoid generating dust. Collect residues and place in sealed plastic bags or other containers for disposal. IF WET: Vacuum/shovel up and place in labelled containers for disposal.</li> <li>ALWAYS: Wash area down with large amounts of water and prevent runoff into drains.</li> <li>If contamination of drains or waterways occurs, advise Emergency Services.</li> </ul>

Personal Protective Equipment advice is contained in Section 8 of the SDS.

# **SECTION 7 Handling and storage**

Safe handling	<ul> <li>Avoid all personal contact, including inhalation.</li> <li>Wear protective clothing when risk of exposure occurs.</li> <li>Use in a well-ventilated area.</li> <li>Prevent concentration in hollows and sumps.</li> <li>DO NOT enter confined spaces until atmosphere has been checked.</li> <li>DO NOT enter confined spaces until atmosphere has been checked.</li> <li>Avoid contact with incompatible materials.</li> <li>When handling, DO NOT est, drink or smoke.</li> <li>Keep containers securely sealed when not in use.</li> <li>Avoid physical damage to containers.</li> <li>Aways wash hands with soap and water after handling.</li> <li>Work clothes should be laundered separately. Launder contaminated clothing before re-use.</li> <li>Use good occupational work practice.</li> <li>Observe manufacturer's storage and handling recommendations contained within this SDS.</li> <li>Atmosphere should be regularly checked against established exposure standards to ensure safe working conditions are maintained.</li> <li>Organic powders when finely divided over a range of concentrations regardless of particulate size or shape and suspended in air or some other oxidizing medium may form explosive dust-air mixtures and result in a fine or dust explosion (including secondary explosion Minimise airborne dust and eliminate all ignition sources. Keep away from heat, hot surfaces, sparks, and flame.</li> <li>Establish good housekeeping practices.</li> <li>Remove dust accumulations on a regular basis by vacuuming or gentle sweeping to avoid creating dust clouds.</li> <li>Use continuous suction at points of dust generation to capture and minimise the accumulation of dusts. Particular attention should be given to overhead and hidden horizontal surfaces to minimise the probability of a "secondary" explosion. According to NFPA Standard 684, dust layers 1732 in (0.8 mn) thick can be sufficient to warrant immediate cleaning of the area.</li> <li>Do not use air hoses for cleaning.</li> <li>Mininnise dry sweeping to avoid generation of dust clo</li></ul>
Other information	<ul> <li>Store in original containers.</li> <li>Keep containers securely sealed.</li> <li>Store in a cool, dry area protected from environmental extremes.</li> <li>Store away from incompatible materials and foodstuff containers.</li> <li>Protect containers against physical damage and check regularly for leaks.</li> <li>Observe manufacturer's storage and handling recommendations contained within this SDS.</li> <li>For major quantities:</li> <li>Consider storage in bunded areas - ensure storage areas are isolated from sources of community water (including stormwater, ground water, lakes and streams).</li> <li>Ensure that accidental discharge to air or water is the subject of a contingency disaster management plan; this may require consultation with local authorities.</li> </ul>

Conditions for safe storage, including any incompatibilities

Suitable container	<ul> <li>Polyethylene or polypropylene container.</li> <li>Check all containers are clearly labelled and free from leaks.</li> </ul>
Storage incompatibility	Avoid reaction with oxidising agents

### **SECTION 8 Exposure controls / personal protection**

### **Control parameters**

Occupational Exposure Limits	(OEL)			
INGREDIENT DATA				
Not Available				
Emergency Limits				
Ingredient	TEEL-1	TEEL-2		TEEL-3
Ranvet's Hoof Food	Not Available	Not Available		Not Available
Ingredient	Original IDLH		Revised IDLH	
methionine	Not Available		Not Available	
Occupational Exposure Bandir	ng			
Ingredient	Occupational Exposure Band Rating		Occupational Expo	sure Band Limit
methionine	E	E		
Notes:	Occupational exposure banding is a process of adverse health outcomes associated with exp to a range of exposure concentrations that are	osure. The output of this	process is an occupatio	

# MATERIAL DATA

It is the goal of the ACGIH (and other Agencies) to recommend TLVs (or their equivalent) for all substances for which there is evidence of health effects at airborne concentrations encountered in the workplace.

At this time no TLV has been established, even though this material may produce adverse health effects (as evidenced in animal experiments or clinical experience). Airborne concentrations must be maintained as low as is practically possible and occupational exposure must be kept to a minimum.

NOTE: The ACGIH occupational exposure standard for Particles Not Otherwise Specified (P.N.O.S) does NOT apply.

Sensory irritants are chemicals that produce temporary and undesirable side-effects on the eyes, nose or throat. Historically occupational exposure standards for these irritants have been based on observation of workers' responses to various airborne concentrations. Present day expectations require that nearly every individual should be protected against even minor sensory irritation and exposure standards are established using uncertainty factors or safety factors of 5 to 10 or more. On occasion animal no-observable-effect-levels (NOEL) are used to determine these limits where human results are unavailable. An additional approach, typically used by the TLV committee (USA) in determining respiratory standards for this group of chemicals, has been to assign ceiling values (TLV C) to rapidly acting irritants and to assign short-term exposure limits (TLV STELs) when the weight of evidence from irritation, bioaccumulation and other endpoints combine to warrant such a limit. In contrast the MAK Commission (Germany) uses a five-category system based on intensive odour, local irritation, and elimination half-life. However this system is being replaced to be consistent with the European Union (EU) Scientific Committee for Occupational Exposure Limits (SCOEL); this is more closely allied to that of the USA.

OSHA (USA) concluded that exposure to sensory irritants can:

- cause inflammation
- cause increased susceptibility to other irritants and infectious agents
- lead to permanent injury or dysfunction
- permit greater absorption of hazardous substances and
- acclimate the worker to the irritant warning properties of these substances thus increasing the risk of overexposure.

None required when handling small quantities

### Exposure controls

	None required when handling small quantities. OTHERWISE: Engineering controls are used to remove a hazard or place a can be highly effective in protecting workers and will typically The basic types of engineering controls are: Process controls which involve changing the way a job activit Enclosure and/or isolation of emission source which keeps a strategically "adds" and "removes" air in the work environment design of a ventilation system must match the particular proc Employers may need to use multiple types of controls to prev • Local exhaust ventilation is required where solids are han certain proportion will be powdered by mutual friction. • If in spite of local exhaust an adverse concentration of th Such protection might consist of: (a): particle dust respirators, if necessary, combined with an a (b): filter respirators with absorption cartridge or canister of th (c): fresh-air hoods or masks. Air contaminants generated in the workplace possess varying circulating air required to effectively remove the contaminant	be independent of worker interactions to provide this by or process is done to reduce the risk. selected hazard "physically" away from the worker ar it. Ventilation can remove or dilute an air contaminant ess and chemical or contaminant in use. rent employee overexposure. Indied as powders or crystals; even when particulates e substance in air could occur, respiratory protection absorption cartridge; ie right type; g "escape" velocities which, in turn, determine the "ca	high level of protection. Ind ventilation that t if designed properly. The are relatively large, a should be considered.		
Appropriate engineering	Type of Contaminant:		Air Speed:		
controls	direct spray, spray painting in shallow booths, drum filling, (active generation into zone of rapid air motion)	conveyer loading, crusher dusts, gas discharge	1-2.5 m/s (200-500 f/min.)		
	grinding, abrasive blasting, tumbling, high speed wheel ger zone of very high rapid air motion).	nerated dusts (released at high initial velocity into	2.5-10 m/s (500-2000 f/min.)		
	Within each range the appropriate value depends on:				
	Lower end of the range	Upper end of the range			
	1: Room air currents minimal or favourable to capture	1: Disturbing room air currents			
	2: Contaminants of low toxicity or of nuisance value only.	2: Contaminants of high toxicity			
	3: Intermittent, low production.	3: High production, heavy use			
	4: Large hood or large air mass in motion	4: Small hood-local control only			
	Simple theory shows that air velocity falls rapidly with distance away from the opening of a simple extraction pipe. Velocity generally decreases with the square of distance from the extraction point (in simple cases). Therefore the air speed at the extraction point should be adjusted, accordingly, after reference to distance from the contaminating source. The air velocity at the extraction fan, for example, should be a minimum of 4-10 m/s (800-2000 f/min) for extraction of crusher dusts generated 2 metres distant from the extraction point. Other mechanical considerations, producing performance deficits within the extraction apparatus, make it essential that theoretical air velocities are multiplied by factors of 10 or more when extraction systems are installed or used.				
Individual protection measures, such as personal protective equipment					
Eye and face protection	<ul> <li>No special equipment for minor exposure i.e. when handling small quantities.</li> <li>OTHERWISE:</li> <li>Safety glasses with side shields.</li> <li>Contact lenses may pose a special hazard; soft contact lenses may absorb and concentrate irritants. A written policy document, describing the wearing of lenses or restrictions on use, should be created for each workplace or task. This should include a review of lens absorption and adsorption for the class of chemicals in use and an account of injury experience. Medical and first-aid personnel should be trained in their removal and suitable equipment should be readily available. In the event of chemical exposure, begin eye irrigation immediately and remove contact lens as soon as practicable. Lens should be removed at the first signs of eye redness or irritation - lens should be removed in a clean environment only after workers have washed hands thoroughly. [CDC NIOSH Current Intelligence Bulletin 59], [AS/NZS 1336 or national equivalent]</li> </ul>				
Skin protection	See Hand protection below				
Hands/feet protection	No special equipment needed when handling small quantities. <b>OTHERWISE</b> : Wear chemical protective gloves, e.g. PVC. Experience indicates that the following polymers are suitable as glove materials for protection against undissolved, dry solids, where abrasive particles are not present. • polychloroprene. • nitrile rubber. • butyl rubber. • fluorocaoutchouc. • polyvinyl chloride. Gloves should be examined for wear and/ or degradation constantly.				
Body protection	See Other protection below				
Other protection	No special equipment needed when handling small quantities	5.			
	I				

OTHERWISE: ► Overalls. ► Barrier cream.

Eyewash unit.

#### **Respiratory protection**

Type -P Filter of sufficient capacity. (AS/NZS 1716 & 1715, EN 143:2000 & 149:2001, ANSI Z88 or national equivalent)

Required Minimum Protection Factor	Half-Face Respirator	Full-Face Respirator	Powered Air Respirator
up to 10 x ES	P1 Air-line*	-	PAPR-P1 -
up to 50 x ES	Air-line**	P2	PAPR-P2
up to 100 x ES	-	P3	-
		Air-line*	-
100+ x ES	-	Air-line**	PAPR-P3

\* - Negative pressure demand \*\* - Continuous flow

A(All classes) = Organic vapours, B AUS or B1 = Acid gasses, B2 = Acid gas or hydrogen cyanide(HCN), B3 = Acid gas or hydrogen cyanide(HCN), E = Sulfur dioxide(SO2), G = Agricultural chemicals, K = Ammonia(NH3), Hg = Mercury, NO = Oxides of nitrogen, MB = Methyl bromide, AX = Low boiling point organic compounds(below 65 degC)

 $\cdot$  Respirators may be necessary when engineering and administrative controls do not adequately prevent exposures.

The decision to use respiratory protection should be based on professional judgment that takes into account toxicity information, exposure measurement data, and frequency and likelihood of the worker's exposure - ensure users are not subject to high thermal loads which may result in heat stress or distress due to personal protective equipment (powered, positive flow, full face apparatus may be an option).

Published occupational exposure limits, where they exist, will assist in determining the adequacy of the selected respiratory protection. These may be government mandated or vendor recommended.

· Certified respirators will be useful for protecting workers from inhalation of particulates when properly selected and fit tested as part of a complete respiratory protection program.

• Where protection from nuisance levels of dusts are desired, use type N95 (US) or type P1 (EN143) dust masks. Use respirators and components tested and approved under appropriate government standards such as NIOSH (US) or CEN (EU)

 $\cdot$  Use approved positive flow mask if significant quantities of dust becomes airborne.

Try to avoid creating dust conditions.

# **SECTION 9** Physical and chemical properties

#### Information on basic physical and chemical properties

Appearance	Beige coloured powder; does not mix with water.		
Physical state	Divided Solid	Relative density (Water = 1)	Not Available
Odour	Not Available	Partition coefficient n-octanol / water	Not Available
Odour threshold	Not Available	Auto-ignition temperature (°C)	Not Available
pH (as supplied)	Not Applicable	Decomposition temperature (°C)	Not Available
Melting point / freezing point (°C)	Not Available	Viscosity (cSt)	Not Available
Initial boiling point and boiling range (°C)	Not Available	Molecular weight (g/mol)	Not Applicable
Flash point (°C)	Not Applicable	Taste	Not Available
Evaporation rate	Not Available	Explosive properties	Not Available
Flammability	Not Applicable	Oxidising properties	Not Available
Upper Explosive Limit (%)	Not Available	Surface Tension (dyn/cm or mN/m)	Not Applicable
Lower Explosive Limit (%)	Not Available	Volatile Component (%vol)	Negligible
Vapour pressure (kPa)	Negligible	Gas group	Not Available
Solubility in water	Immiscible	pH as a solution (1%)	Not Applicable
Vapour density (Air = 1)	Not Available	VOC g/L	Not Available

### **SECTION 10 Stability and reactivity**

Reactivity	See section 7
Chemical stability	<ul> <li>Unstable in the presence of incompatible materials.</li> <li>Product is considered stable.</li> <li>Hazardous polymerisation will not occur.</li> </ul>
Possibility of hazardous reactions	See section 7
Conditions to avoid	See section 7
Incompatible materials	See section 7
Hazardous decomposition products	See section 5

#### **SECTION 11 Toxicological information**

	Persons with impaired respiratory function, airway diseases and condi disability if excessive concentrations of particulate are inhaled.	tions such as emp	hysema or chronic bronchitis, may incur further
Inhaled	If prior damage to the circulatory or nervous systems has occurred or if kidney damage has been sustained, proper screenings should be conducted on individuals who may be exposed to further risk if handling and use of the material result in excessive exposures. Not normally a hazard due to non-volatile nature of product		
Ingestion	Accidental ingestion of the material may be damaging to the health of the individual. Magnesium salts are generally absorbed so slowly that oral administration causes few toxic effects with purging being the most significant. If evacuation fails due to bowel obstruction or atony, mucosal irritation and absorption may result. Side effects of magnesium salts include upset stomach, dry mouth, dry nose and dry throat, drowsiness, nausea, heartburn and thickening of the mucous in the throat and nose. Systemically the magnesium ion produces electrolyte imbalance, central nervous system depression, neurological and cardiac involvement, reflex abolition and death from respiratory paralysis. These effects are rare in the absence of intestinal or renal disorders. Early signs and symptoms of magnesium intoxication include nausea, vomiting, malaise and confusion. Deep tendon reflexes may be diminished. central nervous system depression may progress to coma and paralysis of the release of acetylcholine at myoneuronal junctions. Central nervous system depression may be compounded by depressed function of the respiratory musculature. Hypotension may also ensue as a result of peripheral vasodilation and/ or decreased cardiac output secondary to conduction defects. Bradycardia is common, leading to eventual arrest in diastole.		
Skin Contact	Evidence exists, or practical experience predicts, that the material either produces inflammation of the skin in a substantial number of individuals following direct contact, and/or produces significant inflammation when applied to the healthy intact skin of animals, for up to four hours, such inflammation being present twenty-four hours or more after the end of the exposure period. Skin irritation may also be present after prolonged or repeated exposure; this may result in a form of contact dermatitis (nonallergic). The dermatitis is often characterised by skin redness (erythema) and swelling (oedema) which may progress to blistering (vesiculation), scaling and thickening of the epidermis. At the microscopic level there may be intercellular oedema of the spongy layer of the skin (spongiosis) and intracellular oedema of the epidermis. The material may accentuate any pre-existing dermatitis condition Open cuts, abraded or irritated skin should not be exposed to this material Entry into the blood-stream through, for example, cuts, abrasions, puncture wounds or lesions, may produce systemic injury with harmful effects. Examine the skin prior to the use of the material and ensure that any external damage is suitably protected.		
Eye	Evidence exists, or practical experience predicts, that the material may cause eye irritation in a substantial number of individuals and/or may produce significant ocular lesions which are present twenty-four hours or more after instillation into the eye(s) of experimental animals. Repeated or prolonged eye contact may cause inflammation characterised by temporary redness (similar to windburn) of the conjunctiva (conjunctivitis); temporary impairment of vision and/or other transient eye damage/ulceration may occur.		
Chronic	Repeated or long-term occupational exposure is likely to produce cum A case of chronic abuse of magnesium citrate ( a mild purgative), by a lethargy and severe refractory hypotension. Pathology revealed extrer have a perforated duodenal ulcer. She died after peritoneal dialysis (w A patient with normal kidney function developed symptomatic hyperma grams of magnesium sulfate over 18 hours. When magnesium sulfate was given to pregnant rats, a sharp reduction Long term exposure to high dust concentrations may cause changes in micron penetrating and remaining in the lung. A prime symptom is brea May cause SENSITISATION by skin contact.	62 year-old womane hypermagnesa hich reduced seru Ignesaemia with ro n of both the numina h lung function (i.e	ts involving organs or biochemical systems. In, has been reported. Symptoms of abuse included emia [6.25 mmol per litre]. She also was found to m-magnesium and reduced hypotension. aspiratory arrest and bradycardia after receiving 90 oper and the weight of the offspring was observed. . pneumoconiosis) caused by particles less than 0.5
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Ranvet's Hoof Food	A case of chronic abuse of magnesium citrate ( a mild purgative), by a lethargy and severe refractory hypotension. Pathology revealed extrem have a perforated duodenal ulcer. She died after peritoneal dialysis (w A patient with normal kidney function developed symptomatic hyperma grams of magnesium sulfate over 18 hours. When magnesium sulfate was given to pregnant rats, a sharp reduction Long term exposure to high dust concentrations may cause changes in micron penetrating and remaining in the lung. A prime symptom is breading cause SENSITISATION by skin contact.	62 year-old woma ne hypermagnesa hich reduced seru gnesaemia with ro n of both the numi n lung function (i.e. athlessness. Lung IRRITATION Not Available IRRITATION Eye: no adverse	ts involving organs or biochemical systems. In, has been reported. Symptoms of abuse included emia [6.25 mmol per litre]. She also was found to m-magnesium and reduced hypotension. espiratory arrest and bradycardia after receiving 90 per and the weight of the offspring was observed.
Ranvet's Hoof Food	A case of chronic abuse of magnesium citrate ( a mild purgative), by a lethargy and severe refractory hypotension. Pathology revealed extrem have a perforated duodenal ulcer. She died after peritoneal dialysis (w A patient with normal kidney function developed symptomatic hyperma grams of magnesium sulfate over 18 hours. When magnesium sulfate was given to pregnant rats, a sharp reduction Long term exposure to high dust concentrations may cause changes in micron penetrating and remaining in the lung. A prime symptom is breat May cause SENSITISATION by skin contact.  TOXICITY Not Available TOXICITY Inhalation (Rat) LC50: >5.25 mg/l4h <sup>[1]</sup>	62 year-old woma ne hypermagnesa hich reduced seru gnesaemia with ro n of both the numi n lung function (i.e athlessness. Lung IRRITATION Not Available IRRITATION Eye: no adverse Skin: no adverse toxicity 2. Value of	ts involving organs or biochemical systems. In, has been reported. Symptoms of abuse include- emia [6.25 mmol per litre]. She also was found to m-magnesium and reduced hypotension. aspiratory arrest and bradycardia after receiving 90 oper and the weight of the offspring was observed. . pneumoconiosis) caused by particles less than 0. shadows show on X-ray. e effect observed (not irritating) <sup>[1]</sup> e effect observed (not irritating) <sup>[1]</sup>
Ranvet's Hoof Food methionine	A case of chronic abuse of magnesium citrate ( a mild purgative), by a lethargy and severe refractory hypotension. Pathology revealed extrer have a perforated duodenal ulcer. She died after peritoneal dialysis (w A patient with normal kidney function developed symptomatic hyperma grams of magnesium sulfate over 18 hours. When magnesium sulfate was given to pregnant rats, a sharp reduction Long term exposure to high dust concentrations may cause changes in micron penetrating and remaining in the lung. A prime symptom is breat May cause SENSITISATION by skin contact.  TOXICITY Not Available TOXICITY Inhalation (Rat) LC50: >5.25 mg/l4h <sup>[1]</sup> Oral (Rat) LD50: 36000 mg/kg <sup>[2]</sup> 1. Value obtained from Europe ECHA Registered Substances - Acute	62 year-old woma ne hypermagnesa hich reduced seru gnesaemia with ro n of both the numi n lung function (i.e athlessness. Lung IRRITATION Not Available IRRITATION Eye: no adverse Skin: no adverse toxicity 2. Value of	ts involving organs or biochemical systems. In, has been reported. Symptoms of abuse include emia [6.25 mmol per litre]. She also was found to m-magnesium and reduced hypotension. aspiratory arrest and bradycardia after receiving 90 oper and the weight of the offspring was observed. . pneumoconiosis) caused by particles less than 0. shadows show on X-ray. e effect observed (not irritating) <sup>[1]</sup> e effect observed (not irritating) <sup>[1]</sup>
Ranvet's Hoof Food methionine Legend:	A case of chronic abuse of magnesium citrate ( a mild purgative), by a lethargy and severe refractory hypotension. Pathology revealed extrer have a perforated duodenal ulcer. She died after peritoneal dialysis (w A patient with normal kidney function developed symptomatic hyperma grams of magnesium sulfate over 18 hours. When magnesium sulfate over 18 hours. When magnesium sulfate was given to pregnant rats, a sharp reduction Long term exposure to high dust concentrations may cause changes is micron penetrating and remaining in the lung. A prime symptom is breat May cause SENSITISATION by skin contact.  TOXICITY Not Available  TOXICITY Inhalation (Rat) LC50: >5.25 mg/l4h <sup>[1]</sup> Oral (Rat) LD50: 36000 mg/kg <sup>[2]</sup> 1. Value obtained from Europe ECHA Registered Substances - Acute specified data extracted from RTECS - Register of Toxic Effect of chere. No significant acute toxicological data identified in literature search.	62 year-old woma ne hypermagnesa hich reduced seru gnesaemia with ro n of both the numi n lung function (i.e athlessness. Lung IRRITATION Not Available IRRITATION Eye: no adverse Skin: no adverse toxicity 2. Value of	ts involving organs or biochemical systems. In, has been reported. Symptoms of abuse include- emia [6.25 mmol per litre]. She also was found to m-magnesium and reduced hypotension. aspiratory arrest and bradycardia after receiving 90 oper and the weight of the offspring was observed. . pneumoconiosis) caused by particles less than 0. shadows show on X-ray. e effect observed (not irritating) <sup>[1]</sup> e effect observed (not irritating) <sup>[1]</sup>
Ranvet's Hoof Food methionine <i>Legend:</i> Ranvet's Hoof Food	A case of chronic abuse of magnesium citrate ( a mild purgative), by a lethargy and severe refractory hypotension. Pathology revealed extrer have a perforated duodenal ulcer. She died after peritoneal dialysis (w A patient with normal kidney function developed symptomatic hypermagrams of magnesium sulfate over 18 hours. When magnesium sulfate over 18 hours. When magnesium sulfate was given to pregnant rats, a sharp reduction Long term exposure to high dust concentrations may cause changes in micron penetrating and remaining in the lung. A prime symptom is breat May cause SENSITISATION by skin contact. <b>TOXICITY</b> Inhalation (Rat) LC50: >5.25 mg/l4h <sup>[1]</sup> Oral (Rat) LD50: 36000 mg/kg <sup>[2]</sup> 1. Value obtained from Europe ECHA Registered Substances - Acute specified data extracted from RTECS - Register of Toxic Effect of cheir No significant acute toxicological data identified in literature search.	62 year-old woma ne hypermagnesa hich reduced seru gnesaemia with ro n of both the numi n lung function (i.e. athlessness. Lung IRRITATION Not Available IRRITATION Eye: no adverse Skin: no adverse toxicity 2. Value of mical Substances	ts involving organs or biochemical systems. In, has been reported. Symptoms of abuse include emia [6.25 mmol per litre]. She also was found to m-magnesium and reduced hypotension. espiratory arrest and bradycardia after receiving 90 ber and the weight of the offspring was observed. . pneumoconiosis) caused by particles less than 0. shadows show on X-ray. e effect observed (not irritating) <sup>[1]</sup> e effect observed (not irritating) <sup>[1]</sup> batained from manufacturer's SDS. Unless otherwise
Ranvet's Hoof Food methionine <i>Legend:</i> Ranvet's Hoof Food Acute Toxicity	A case of chronic abuse of magnesium citrate ( a mild purgative), by a lethargy and severe refractory hypotension. Pathology revealed extrem have a perforated duodenal ulcer. She died after peritoneal dialysis (w A patient with normal kidney function developed symptomatic hypermagrams of magnesium sulfate over 18 hours. When magnesium sulfate was given to pregnant rats, a sharp reduction Long term exposure to high dust concentrations may cause changes in micron penetrating and remaining in the lung. A prime symptom is breat May cause SENSITISATION by skin contact.  TOXICITY Not Available  TOXICITY Inhalation (Rat) LC50: >5.25 mg/l4h <sup>[1]</sup> Oral (Rat) LD50: 36000 mg/kg <sup>[2]</sup> 1. Value obtained from Europe ECHA Registered Substances - Acute specified data extracted from RTECS - Register of Toxic Effect of chert	62 year-old woma ne hypermagnesa hich reduced seru gnesaemia with re n of both the numi n lung function (i.e. athlessness. Lung IRRITATION Not Available IRRITATION Eye: no adverse Skin: no adverse toxicity 2. Value of mical Substances	e effect observed (not irritating) <sup>[1]</sup> e effect observed (not irritating) <sup>[1]</sup>
Ranvet's Hoof Food methionine <i>Legend:</i> Ranvet's Hoof Food Skin Irritation/Corrosion Serious Eye	A case of chronic abuse of magnesium citrate ( a mild purgative), by a lethargy and severe refractory hypotension. Pathology revealed extrem have a perforated duodenal ulcer. She died after peritoneal dialysis (w A patient with normal kidney function developed symptomatic hypermagrams of magnesium sulfate over 18 hours. When magnesium sulfate was given to pregnant rats, a sharp reduction Long term exposure to high dust concentrations may cause changes in micron penetrating and remaining in the lung. A prime symptom is breat May cause SENSITISATION by skin contact.  TOXICITY Not Available TOXICITY Inhalation (Rat) LC50: >5.25 mg/l4h <sup>[1]</sup> Oral (Rat) LD50: 36000 mg/kg <sup>[2]</sup> 1. Value obtained from Europe ECHA Registered Substances - Acute specified data extracted from RTECS - Register of Toxic Effect of chert	62 year-old womane hypermagnesa hich reduced seru sgnesaemia with room of both the numina lung function (i.e. athlessness. Lung IRRITATION Not Available IRRITATION Eye: no adverse Skin: no adverse toxicity 2. Value of nical Substances Carcinogenicity Reproductivity	effect observed (not irritating) <sup>[1]</sup> effect observed (not irritating) <sup>[1]</sup>

# **SECTION 12 Ecological information**

Toxicity Endpoint Test Duration (hr) Species Value Source Ranvet's Hoof Food Not Not Not Not Available Not Available Available Available Available methionine Source Endpoint Test Duration (hr) Species Value 72h >1000mg/l EC50 Algae or other aquatic plants 2 48h Crustacea 2 EC50 324mg/l

	EC50(ECx)	48h	Crustacea	324mg/l	2
	LC50	96h	Fish	>3200mg/l	2
Legend:	Ecotox databas	, , ,	red Substances - Ecotoxicological Information - A zard Assessment Data 6. NITE (Japan) - Bioconce		,

for magnesium compounds in general:

Fish LC50: 100-400 mg/l

### DO NOT discharge into sewer or waterways.

# Persistence and degradability

Ingredient	Persistence: Water/Soil	Persistence: Air
methionine	LOW	LOW
Bioaccumulative potential		
Ingredient	Bioaccumulation	
methionine	LOW (LogKOW = -1.87)	
Mobility in soil		
Ingredient	Mobility	
methionine	LOW (Log KOC = 9.356)	

# **SECTION 13 Disposal considerations**

Waste treatment methods	
Product / Packaging disposal	<ul> <li>Legislation addressing waste disposal requirements may differ by country, state and/ or territory. Each user must refer to laws operating in their area. In some areas, certain wastes must be tracked.</li> <li>A Hierarchy of Controls seems to be common - the user should investigate: <ul> <li>Reduction</li> <li>Reuse</li> <li>Recycling</li> <li>Disposal (if all else fails)</li> </ul> </li> <li>This material may be recycled if unused, or if it has not been contaminated so as to make it unsuitable for its intended use. Shelf life considerations should also be applied in making decisions of this type. Note that properties of a material may change in use, and recycling or reuse may not always be appropriate. In most instances the supplier of the material should be consulted.</li> <li><b>DO NOT</b> allow wash water from cleaning or process equipment to enter drains.</li> <li>It may be necessary to collect all wash water for treatment before disposal.</li> <li>In all cases disposal to sever may be subject to local laws and regulations and these should be considered first.</li> <li>Where in doubt contact the responsible authority.</li> <li>Recycle wherever possible.</li> <li>Consult manufacturer for recycling options or consult local or regional waste management authority for disposal if no suitable treatment or disposal facility can be identified.</li> <li>Dispose of by: burial in a land-fill specifically licensed to accept chemical and / or pharmaceutical wastes or Incineration in a licensed apparatus (after admixture with suitable combustible material)</li> <li>Decontaminate empty containers. Observe all label safeguards until containers are cleaned and destroyed.</li> </ul>

# **SECTION 14 Transport information**

Labels Required		
Marine Pollutant	NO	
HAZCHEM	Not Applicable	

Land transport (ADG): NOT REGULATED FOR TRANSPORT OF DANGEROUS GOODS

Air transport (ICAO-IATA / DGR): NOT REGULATED FOR TRANSPORT OF DANGEROUS GOODS

Sea transport (IMDG-Code / GGVSee): NOT REGULATED FOR TRANSPORT OF DANGEROUS GOODS

14.7.1. Transport in bulk according to Annex II of MARPOL and the IBC code Not Applicable

# 14.7.2. Transport in bulk in accordance with MARPOL Annex V and the IMSBC Code

Product name	Group
methionine	Not Available

### 14.7.3. Transport in bulk in accordance with the IGC Code

Product name	Ship Type
methionine	Not Available

# **SECTION 15 Regulatory information**

Safety, health and environmental regulations / legislation specific for the substance or mixture

methionine is found on the following regulatory lists

Australian Inventory of Industrial Chemicals (AIIC)

### Additional Regulatory Information Not Applicable

# **National Inventory Status**

National Inventory	Status		
Australia - AIIC / Australia Non- Industrial Use	Yes		
Canada - DSL	Yes		
Canada - NDSL	Yes		
China - IECSC	Yes		
Europe - EINEC / ELINCS / NLP	Yes		
Japan - ENCS	Yes		
Korea - KECI	Yes		
New Zealand - NZIoC	Yes		
Philippines - PICCS	Yes		
USA - TSCA	Yes		
Taiwan - TCSI	Yes		
Mexico - INSQ	Yes		
Vietnam - NCI	Yes		
Russia - FBEPH	Yes		
Legend:	Yes = All CAS declared ingredients are on the inventory No = One or more of the CAS listed ingredients are not on the inventory. These ingredients may be exempt or will require registration.		

# **SECTION 16 Other information**

Revision Date	03/10/2023
Initial Date	11/08/2012

#### SDS Version Summarv

Version	Date of Update	Sections Updated
6.1	12/23/2022	Classification review due to GHS Revision change.
7.1	03/10/2023	Classification change due to full database hazard calculation/update.

#### Other information

Classification of the preparation and its individual components has drawn on official and authoritative sources as well as independent review by the Chemwatch Classification committee using available literature references.

The SDS is a Hazard Communication tool and should be used to assist in the Risk Assessment. Many factors determine whether the reported Hazards are Risks in the workplace or other settings. Risks may be determined by reference to Exposures Scenarios. Scale of use, frequency of use and current or available engineering controls must be considered.

#### Definitions and abbreviations

- PC TWA: Permissible Concentration-Time Weighted Average
- PC STEL: Permissible Concentration-Short Term Exposure Limit
- IARC: International Agency for Research on Cancer
- ACGIH: American Conference of Governmental Industrial Hygienists
- STEL: Short Term Exposure Limit
   TEEL: Temporary Emergency Exposure Limit.
- IDLH: Immediately Dangerous to Life or Health Concentrations
- ES: Exposure Standard
- OSF: Odour Safety Factor
- NOAEL: No Observed Adverse Effect Level LOAEL: Lowest Observed Adverse Effect Level
- TLV: Threshold Limit Value
- LOD: Limit Of Detection
- OTV: Odour Threshold Value
- BCF: BioConcentration Factors
- BEI: Biological Exposure Index
- DNEL: Derived No-Effect Level
- PNEC: Predicted no-effect concentration
- AIIC: Australian Inventory of Industrial Chemicals
- DSL: Domestic Substances List
- NDSL: Non-Domestic Substances List IECSC: Inventory of Existing Chemical Substance in China
- EINECS: European INventory of Existing Commercial chemical Substances
- ELINCS: European List of Notified Chemical Substances
- NLP: No-Longer Polymers
- ENCS: Existing and New Chemical Substances Inventory
   KECI: Korea Existing Chemicals Inventory
   NZIoC: New Zealand Inventory of Chemicals

- PICCS: Philippine Inventory of Chemicals and Chemical Substances
- TSCA: Toxic Substances Control Act
- TCSI: Taiwan Chemical Substance Inventory
- INSQ: Inventario Nacional de Sustancias Químicas

- NCI: National Chemical Inventory
   FBEPH: Russian Register of Potentially Hazardous Chemical and Biological Substances

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